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## Rapid kinetic analyses of the $\text{Na}^+/\text{K}^+$ -ATPase distinguish among different criteria for conformational change

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The  $\text{Na}^+/\text{K}^+$ -ATPase couples the hydrolysis of ATP to the transport of  $\text{Na}^+$  and  $\text{K}^+$  via a phosphorylated intermediate and conformational changes. In order to identify these conformational changes, we have probed the sequence of steps from  $\text{EP}(3\text{Na}_{\text{in}}^+)$  to  $\text{EP} + 3\text{Na}_{\text{out}}^+$  with three fluorescent probes (IAF: 5-iodoacetamidofluorescein; BIPM: *N*-[*p*-(2-benzimidazolyl)phenyl]maleimide; and RH421) and the sensitivity of their fluorescence changes to oligomycin and divalent cations ( $\text{Ca}^{2+}$  and  $\text{Mn}^{2+}$ ). The magnitude ( $\%\Delta F$ ) and rate constant ( $k_{\text{obs}}$ ) of ATP-induced fluorescence changes were measured on a fluorescence stopped-flow apparatus, and yielded the following results. (a) With RH421,  $k_{\text{obs}}$  and  $\%\Delta F$  varied with  $[\text{Na}^+]$  (maximal  $k_{\text{obs}} = 100 \text{ s}^{-1}$ ,  $K_{1/2} = 6 \text{ mM}$ ;  $\%\Delta F_{\text{max}} = 6\%$ ,  $K_{1/2} = 1 \text{ mM}$ ); these values are comparable to those previously reported using IAF-labeled enzyme (Pratap, P.R., Robinson, J.D. and Steinberg, M.I. (1991) *Biochim. Biophys. Acta* 1069, 288–298). (b) With BIPM-labeled enzyme,  $k_{\text{obs}}$  did not vary with  $[\text{Na}^+]$  over the range tested, and was twice as high as the maximum  $k_{\text{obs}}$  for RH421. (c) Treatment with oligomycin reduced  $k_{\text{obs}}$  for all three probes to about the same level ( $\approx 1\text{--}2 \text{ s}^{-1}$ ) while  $\%\Delta F_{\text{max}}$  was largely unaffected. (d) Replacing  $\text{Mg}^{2+}$  with  $\text{Ca}^{2+}$  had similar effects to treatment with oligomycin. (e) RH421 fluorescence change was completely abolished in the presence of oligomycin and  $\text{Ca}^{2+}$ . (f) Replacing  $\text{Mg}^{2+}$  with  $\text{Mn}^{2+}$  decreased IAF fluorescence, i.e., put the enzyme in an  $\text{E}_2$ -like form, reduced  $k_{\text{obs}}$ , and rendered oligomycin less effective in reducing  $k_{\text{obs}}$ . From these results, we conclude: (a) the release of the second/third  $\text{Na}^+$  is the rate-limiting step for the conformational change measured by IAF and charge transfer measured with RH421; (b) BIPM indicates an earlier step, either the deocclusion of  $\text{Na}^+$  and/or the release of the first  $\text{Na}^+$ ; (c) oligomycin blocks  $\text{Na}^+$  deocclusion, and this step is sensitive to the divalent cation used to activate enzyme phosphorylation; and (d)  $\text{Ca}^{2+}$  slows the step reported by IAF as well. These experiments indicate that a simple model with two conformations ( $\text{E}_1$  and  $\text{E}_2$ ) is insufficient to explain transient kinetic data.

### Introduction

The  $\text{Na}^+/\text{K}^+$ -ATPase is the enzymatic equivalent of the sodium pump, utilizing energy derived from the hydrolysis of ATP to pump  $\text{Na}^+$  and  $\text{K}^+$  against their electrochemical gradients (for reviews, see Refs. 1–4). A fundamental question in the study of this pump is: how does the enzyme couple ATP hydrolysis with ion transport?

The standard model is the Albers-Post scheme [1–3]: the enzyme is thought to exist in two conformations,  $\text{E}_1$  and  $\text{E}_2$ , each of which can be either phosphorylated or unphosphorylated, in the sequence  $\text{E}_1 \rightarrow \text{E}_1\text{P} \rightarrow \text{E}_2\text{P} \rightarrow \text{E}_2 \rightarrow \text{E}_1$ , with  $\text{Na}^+$ -activated phosphorylation of  $\text{E}_1$ , and  $\text{K}^+$ -activated hydrolysis of  $\text{E}_2\text{P}$ . Evidence for these two conformations include differences in: (a) affinity for  $\text{Na}^+$  and  $\text{K}^+$  [1–3], (b) tryptic digestion patterns [1,5], (c) fluorescence quantum affinities of intrinsic [1,6] and extrinsic probes [1,7–10], and (d) sensitivity to dephosphorylation by ADP ( $\text{E}_1\text{P}$ ) and by  $\text{K}^+$ -stimulated hydrolysis ( $\text{E}_2\text{P}$ ) [1,3,4,11].

Because of the distinct differences in affinity for  $\text{Na}^+$  and  $\text{K}^+$ , ion transport was assumed to be coupled to conformational changes between  $\text{E}_1$  and  $\text{E}_2$  conformations, defined in terms of the properties listed earlier. These properties were assumed to change together with the conformations at the same points in the en-

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Abbreviations: BIPM, *N*-[*p*-(2-benzimidazolyl)phenyl]maleimide; DCCD, dicyclohexylcarbodiimide;  $\text{E}_1$  and  $\text{E}_2$ , two conformations of  $\text{Na}^+/\text{K}^+$ -ATPase;  $\text{E}_1\text{P}$  and  $\text{E}_2\text{P}$ , phosphorylated forms of  $\text{E}_1$  and  $\text{E}_2$ ; IAF, 5-iodoacetamidofluorescein;  $\text{Na}^+/\text{K}^+$ -ATPase, sodium plus potassium-dependent adenosinetriphosphatase (EC 3.6.1.3).

zyme cycle. Among the discrepancies with this model were measurements of the EP forms dephosphorylated by ADP ( $E_1P$ ) and by  $K^+$  ( $E_2P$ ) that summed to more than 100% of total EP [12–16]. Nørby et al. [12], studying ADP- and  $K^+$ -activated dephosphorylation as a function of  $Na^+$  concentration, showed that at least three successive pools of EP, designated A, B, and C, were required for quantitative description of their data; they explicitly specified that transport preceded the conformational change. Yoda and Yoda [13–16] proposed a similar model with three phosphoenzymes,  $E_1P$ ,  $E^*P$ , and  $E_2P$ , but with the stipulation that  $E^*P$  would be dephosphorylated by both ADP- and  $K^+$ -activated processes; however, the ability of their formulation to account quantitatively for the data of Nørby et al. was not examined. Yoda and Yoda specified that one  $Na^+$  was transported during the  $E_1P$  to  $E^*P$  step, that  $E^*P$  was an  $E_1$  conformation, and that the remaining two  $Na^+$  were transported during the  $E^*P$  to  $E_2P$  step(s), possibly preceding, accompanying, or following the actual conformational change. This conclusion was thermodynamically unsatisfactory: in the presence of ADP,  $E^*P$  may be dephosphorylated back to  $E_1$ , re-forming ATP and thus resulting in the transport of 1  $Na^+$  with no expense in energy.

In both those schemes, transport was at least partially dissociated from the conformational change. Skou [17] presented a more detailed scheme, completely dissociating transport from the conformational change and redefining the  $E_1$  and  $E_2$  conformations in terms of the number of cation binding sites on each, with  $E_1$  having three and  $E_2$  two. This implied that the conformational change must occur after transport of each cation species. Earlier work from this laboratory [18] was consistent with this model: those data indicated that the fluorescence of IAF-labeled enzyme changed at or after the release of the second and/or third  $Na^+$  from the enzyme to the extracellular side; however, this conclusion was based on equating the conformational change to the fluorescence change in the IAF-enzyme.

This paper aims to test the underlying assumption in the Albers-Post model that a single conformational change mediates in the transport of  $Na^+$  across the membrane. We have used enzyme labeled with extrinsic probes of conformational change (BIPM and IAF) and of charge transfer across the membrane (RH421). In addition, we have probed the enzyme cycle with oligomycin, an inhibitor of ATPase activity that favors the  $Na^+$ -sensitive conformation; we have also examined the effects of divalent cations on the fluorescence signals. From these data, we put forward a scheme for the transition mediating  $Na^+$  transport that departs from the traditional concept of two conformations ( $E_1$  and  $E_2$ ): we propose that each step in the transition involves a conformational change, and that the various

indicators of conformational change examined here indicate different steps in the transition.

## Materials and methods

Frozen dog kidneys were obtained from Pel-Freez Biologicals (Rogers, AR), IAF and RH421 from Molecular Probes (Eugene, OR), and BIPM from Wako Chemicals USA (Dallas, TX). Low-fluorescence imidazole, EDTA and ATP were from Sigma (St. Louis, MO); reagent-grade salts were either from Sigma or from Mallinckrodt (Paris, KY).

$Na^+/K^+$ -ATPase was isolated from the outer medulla of the kidney by a modification of method C of Jørgensen [19]; specific activity was in the range of 10–25  $\mu\text{mol } P_i(\text{mg protein})^{-1} \text{ min}^{-1}$  at 37°C (except for this assay, all experiments described in this paper were done at 24–25°C). Enzyme was labelled with IAF as described earlier [10,20] and with BIPM as described by Taniguchi and co-workers [21].

For experiments with RH421, IAF enzyme in the appropriate buffer was mixed with RH421 in ethanol; the final RH421 concentration was 2  $\mu\text{M}$ .

Steady-state experiments were performed as described before [18] on a Perkin-Elmer fluorescence spectrophotometer (model MPF-66; Perkin-Elmer, Oak Brook, IL). Stopped-flow experiments were performed on a stopped-flow fluorimeter (Kinetic Instruments, Ann Arbor, MI) interfaced with a Macintosh IIcx personal computer (Apple Computer, Cupertino, CA) through a MacADIOS interface board (GW Instruments, Somerville, MA), as described earlier [18,22]. For IAF-enzyme, fluorescence excitation wavelength ( $\lambda_{\text{ex}}$ ) was 492 nm; emission was measured with a photomultiplier tube using a cutoff filter (Corning 3-69) with the wavelength of 50% transmission ( $\lambda_{1/2}$ ) = 529 nm. Similarly, for BIPM-enzyme,  $\lambda_{\text{ex}}$  = 308, cutoff filter was Corning O-52 ( $\lambda_{1/2}$  = 360 nm); for RH421,  $\lambda_{\text{ex}}$  = 530 nm, cutoff filter was Corning 2-58 ( $\lambda_{1/2}$  = 646 nm).

In experiments where oligomycin was used, oligomycin in ethanol was added to the enzyme at the time when the enzyme suspension was prepared. An equal volume of ethanol was added to the control enzyme (–oligomycin). This resulted in the enzyme being incubated with oligomycin for 2–4 h prior to the experiment.

Concentrations of free  $Mg^{2+}$  and  $MgATP$  (in Fig. 6) were calculated using dissociation constants presented by Fabiato and Fabiato [23].

## Results

### Kinetics of charge transfer

RH421 belongs to a class of styryl dyes that were developed as reporters of membrane potential changes

[24]. These dyes are believed to work via an electrochromic mechanism: the fluorescence quantum efficiency of these dyes is sensitive to changes in electric field. Recently, Klodos and Forbush [25], Bühler et al. [26], and Stürmer et al. [27] reported ATP-induced fluorescence changes in enzyme (in broken membrane preparations) labelled with RH421 and the related dye RH160. These fluorescence changes were interpreted by Bühler et al. [26] and by Stürmer et al. [27] as indications of charge transport through the  $\text{Na}^+/\text{K}^+$ -ATPase. As charge moves into and out of the membrane during the enzyme cycle, the dye was thought to sense changes in the electric field within the membrane.

In order to compare the kinetics of charge transfer (as measured by RH421) with that of conformational change (as reported by IAF [18]), ATP-induced RH421 fluorescence changes (Fig. 1) were measured as a function of sodium concentration (Fig. 2) under conditions identical to those previously used for measuring conformational changes with IAF, including the use here of IAF-labeled enzyme [18].

The rate constant of the fluorescence change ( $k_{\text{obs}}$ ) rose from  $1 \text{ s}^{-1}$  at  $1 \text{ mM Na}^+$  to about  $100 \text{ s}^{-1}$  at  $155 \text{ mM Na}^+$  (Fig. 2A). These data (Fig. 2A) were fitted with the Hill equation, which yielded a  $K_{1/2}$  of  $6 \text{ mM}$  and a Hill coefficient of  $1.7$ . The saturation value for  $k_{\text{obs}}$  ( $100 \text{ s}^{-1}$ ) was (consistently) about 25% higher than that observed with the IAF enzyme under the same conditions [18], but the values for  $K_{1/2}$  was comparable to that for the IAF fluorescence change [18]. The

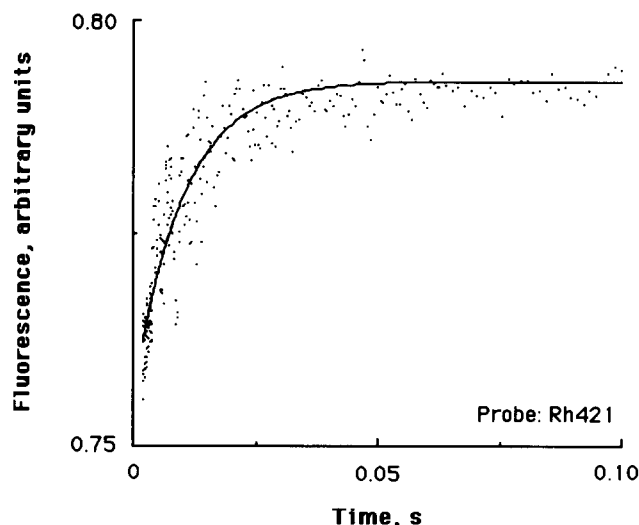


Fig. 1. Stopped-flow response of RH421-enzyme to ATP. Syringe 1:  $100 \mu\text{g/ml}$  enzyme with  $2 \mu\text{M}$  RH421 in buffer ( $50 \text{ mM NaCl}$ ,  $105 \text{ mM choline chloride}$ ,  $4 \text{ mM MgCl}_2$ ,  $1 \text{ mM EDTA}$ ,  $25 \text{ mM imidazole-HCl}$  ( $\text{pH } 7.0$ )). Syringe 2:  $1 \text{ mM ATP}$  in the same buffer. Approx.  $90 \mu\text{l}$  from each syringe was rapidly mixed, and the fluorescence was followed as a function of time ( $t = 24^\circ\text{C}$ ). The solid line is a fit of the data ( $2 \text{ ms}$  to  $1 \text{ s}$ ) by a single exponential. This fit yielded a  $k_{\text{obs}}$  of  $98 \text{ s}^{-1}$  and a  $\% \Delta F$  of  $4.9$ .

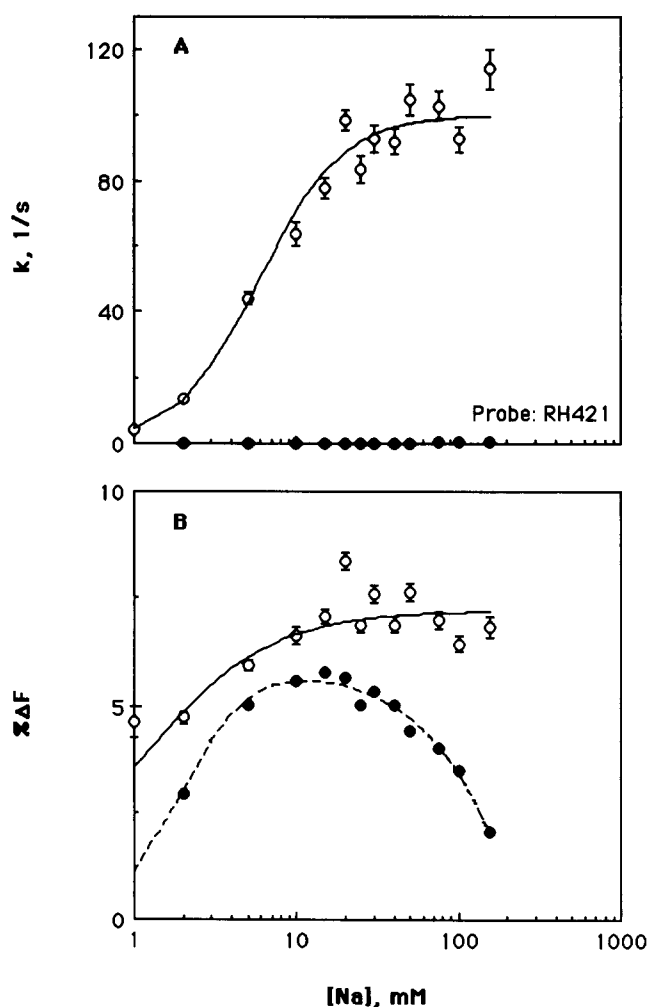


Fig. 2. Rate constant (A) and magnitude (B) of RH421 fluorescence change as a function of  $[\text{Na}^+]$  in the absence ( $\circ$ ) and presence ( $\bullet$ ) of oligomycin (when present added only in the syringe with the enzyme, at a concentration of  $10 \mu\text{g/ml}$ ). Ionic strength was maintained constant with choline chloride. Error bars represent the standard error of the mean. In panel A, the line represents a fit of  $k_{\text{obs}}$  ( $-$  oligomycin) to the Hill equation ( $k_{\text{max}} = 100 \pm 2 \text{ s}^{-1}$ ,  $K_{1/2} = 6.0 \pm 0.3 \text{ mM}$ , and  $n_H = 1.70 \pm 0.07$ ). In panel B, the solid line represents a fit of  $\% \Delta F$  ( $-$  oligomycin) to the Hill equation ( $\% \Delta F_{\text{max}} = 7.2 \pm 0.1$ ,  $K_{1/2} = 1.0 \pm 0.1 \text{ mM}$ , and  $n_H = 1.1 \pm 0.2$ ). For  $\% \Delta F$  ( $+$  oligomycin), the rising portion was fitted with a Hill equation ( $\% \Delta F_{\text{max}} = 6.03 \pm 0.03$ ,  $K_{1/2} = 2.02 \pm 0.03 \text{ mM}$ ,  $n_H = 2.1 \pm 0.1$ ), and the decrease at higher  $\text{Na}^+$  was simultaneously fitted with a straight line (slope =  $-0.0261 \pm 0.0004$ ).

Hill coefficient was higher than that seen with the IAF enzyme, indicating that the RH421 signal may be due to the release of the second and third  $\text{Na}^+$  from the extracellular side.

Apell and co-workers [26,27] have shown that the disappearance of a positive charge (or the appearance of a negative charge) within membrane labeled with RH421 results in the increase of RH421 fluorescence. They interpreted this fluorescence change as a reflection of charge transported across the membrane. Consistent with this interpretation, the magnitude of the

fluorescence change ( $\% \Delta F$ ) represents the steady-state distribution of states with different quantum efficiencies (Fig. 2B), reflecting EP forms prior to charge transfer (low fluorescence) and EP forms after charge is transferred across the membrane (high fluorescence). This parameter also changed with  $\text{Na}^+$  concentration, but the  $K_{1/2}$  was significantly lower, 1 mM; this  $K_{1/2}$  was again comparable to that of the IAF fluorescence change [18].

In the presence of oligomycin,  $k_{\text{obs}}$  was reduced by two orders of magnitude at 155 mM  $\text{Na}^+$ , and no  $\text{Na}^+$  dependence could be detected within the range of concentration used. However,  $\% \Delta F$  still varied with  $\text{Na}^+$ , increasing from 2.9% at 2 mM  $[\text{Na}^+]$  to 5.8% at 20 mM  $[\text{Na}^+]$  (Fig. 2). Thereafter, it decreased, so that at 155 mM  $\text{Na}^+$   $\% \Delta F$  was approximately the same as at 1 mM  $\text{Na}^+$  (Fig. 2B). Such a decrease was not observed with IAF fluorescence over this  $\text{Na}^+$  concen-

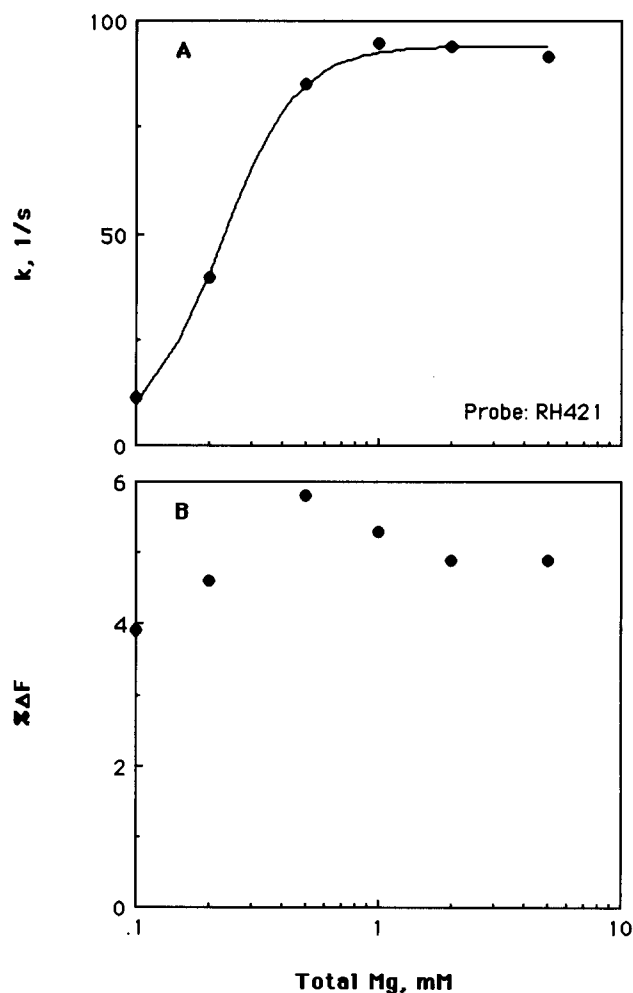


Fig. 3. Rate constant (A) and magnitude (B) of RH421-enzyme fluorescent change as a function of total  $[\text{MgCl}_2]$ . All experimental conditions are as in Fig. 1, except that the EDTA concentration was reduced to 0.1 mM and the  $[\text{MgCl}_2]$  changed as indicated. The solid line in A represents a fit to the Hill equation, with parameters:  $k_{\text{max}} = 94 \pm 1 \text{ s}^{-1}$ ,  $K_{1/2} = 0.221 \pm 0.008 \text{ mM}$ , and  $n_H = 2.7 \pm 0.3$ .

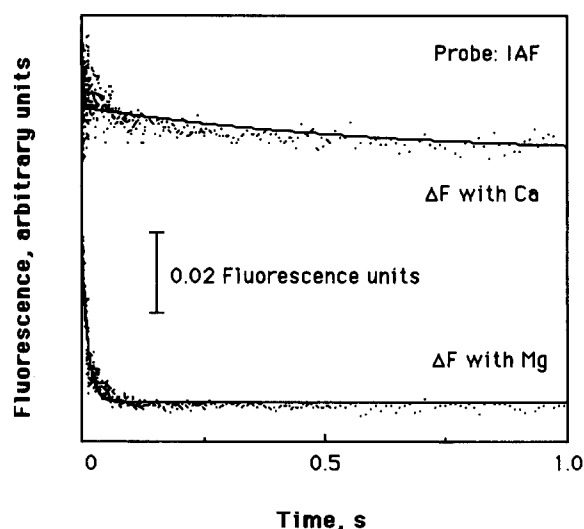


Fig. 4. Stopped-flow fluorescence response of IAF-enzyme with  $\text{Mg}^{2+}$  (lower trace) and with  $\text{Ca}^{2+}$  (upper trace). Syringe 1: 100  $\mu\text{g}/\text{ml}$  IAF-enzyme in buffer (25 mM imidazole-HCl (pH 7.0), 1 mM EDTA, 4 mM divalent cation, 20 mM NaCl, 135 mM choline-Cl); syringe 2, 1 mM ATP in the same buffer. After rapid mixing of 90  $\mu\text{l}$  from each syringe, fluorescence was recorded for 16 s. Only the first second of the fluorescence change is shown here, to show the rapid fluorescence change observed. In order to separate further the points associated with the two traces, the upper trace has been shifted up by 0.04 fluorescence units. Solid lines represent fits of data (2 ms to 16 s) to a single exponential: for enzyme with  $\text{Mg}^{2+}$ ,  $k_{\text{obs}} = 82 \text{ s}^{-1}$ ,  $\% \Delta F = 5.9$ ; for enzyme with  $\text{Ca}^{2+}$ ,  $k_{\text{obs}} = 1.5 \text{ s}^{-1}$ ,  $\% \Delta F = 2.8$ . (In experiments with  $\text{Ca}^{2+}$  there is considerable scatter of the data during the first 50 ms, above and below the fitted line; however, the fit over the full 16 s was better with a monoexponential than a biexponential curve).

tration range (1–155 mM) [18]. The maximal  $\% \Delta F$  was only about 20% lower than in the absence of oligomycin (6% vs. 7.2%).

In order to estimate a lower limit for the  $K_{1/2}$  for the fluorescence increase, these data were fitted to a Hill equation for the rising phase plus a straight line for the declining phase of the curve (the straight line introduced ad hoc to account for the decrease in fluorescence at higher  $[\text{Na}^+]$ ). This fit yielded a lower limit for  $K_{1/2}$  of 2 mM (Fig. 2).

#### Effects of $\text{Mg}^{2+}$

In the presence of 1 mM ATP and 0.1 mM EDTA, RH421 fluorescence changes (Fig. 3) varied with the total  $\text{Mg}^{2+}$  concentration,  $k_{\text{obs}}$  increasing by a factor of approx. 9 from 0.1 to 5 mM  $[\text{Mg}^{2+}]$ . The  $K_{1/2}$  for this increase was 0.22 mM total  $\text{Mg}^{2+}$ . Over this range,  $\% \Delta F$  was between 4 and 6%.

When expressed in terms of free  $[\text{Mg}^{2+}]$  or  $[\text{MgATP}]$ , the data were satisfactorily fitted with a Michaelis-Menten equation (when fitted with a Hill equation, the resultant Hill coefficient was not significantly different from 1). For  $k_{\text{obs}}$  vs.  $[\text{free } \text{Mg}^{2+}]$ , the maximal  $k_{\text{obs}}$  was  $95 \pm 3 \text{ s}^{-1}$ , with a  $K_{1/2}$  of  $13.2 \pm 3.3$

$\mu\text{M}$ . For  $k_{\text{obs}}$  as a function of  $[\text{MgATP}]$ , the fits yielded a maximal  $k_{\text{obs}}$  of  $132 \pm 15$ , with a  $K_{1/2}$  of  $200 \pm 70$   $\mu\text{M}$ .

### Effects of $\text{Ca}^{2+}$

The  $\text{Na}^+/\text{K}^+$ -ATPase requires a divalent cation for activity. The optimal cation is  $\text{Mg}^{2+}$ , but the enzyme can hydrolyze ATP in the presence of a range of alternative divalent cations [28]. When  $\text{Mg}^{2+}$  is replaced by  $\text{Ca}^{2+}$ , the enzyme is phosphorylated very slowly [29], while ATP hydrolysis is almost abolished.

When  $\text{Mg}^{2+}$  was replaced by  $\text{Ca}^{2+}$ , the IAF-enzyme showed a much reduced fluorescence change (Fig. 4):  $k_{\text{obs}}$  was 1.9% that with  $\text{Mg}^{2+}$ , and  $\%\Delta F$  was about half that with  $\text{Mg}^{2+}$ .

These results could mean either (a)  $\text{Ca}^{2+}$  slowed the step reported by IAF so that it now became the rate-limiting step (thus affecting both  $k_{\text{obs}}$  and  $\%\Delta F$ ), or (b)  $\text{Ca}^{2+}$  made an earlier step rate-limiting and simultaneously affected the step reported by IAF. If possibility (a) were true, then  $\text{Ca}^{2+}$  would have no effect on the RH421 fluorescence change, since this occurs prior to the IAF signal. If (b) were true,  $\text{Ca}^{2+}$  would slow  $k_{\text{obs}}$  for the RH421 signal, while not affecting  $\%\Delta F$ .

In order to test these two possibilities, RH421 fluorescence changes were measured with  $\text{Ca}^{2+}$ . In these experiments,  $k_{\text{obs}}$  was reduced by two orders of magnitude (Fig. 5, upper trace), to approximately the same level as with IAF-enzyme (Fig. 4). But unlike experiments with the IAF fluorescence, the  $\%\Delta F$  was essentially the same as with  $\text{Mg}^{2+}$ . The RH421 response

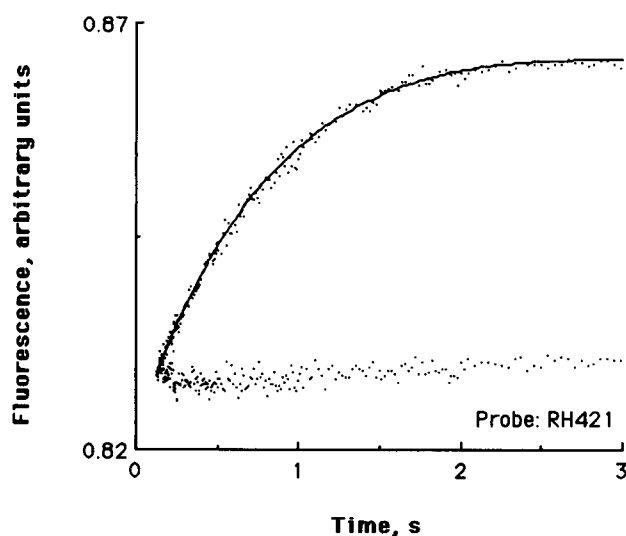


Fig. 5. Stopped-flow fluorescence response of RH421-enzyme when  $\text{Mg}^{2+}$  is replaced with  $\text{Ca}^{2+}$ , in the absence (upper trace) and presence (lower trace, dots) of oligomycin. Experimental conditions were the same as in Fig. 2, except that: syringe 1 contains, in addition, 2  $\mu\text{M}$  RH421 and 4 mM  $\text{CaCl}_2$  was substituted for 4 mM  $\text{MgCl}_2$  for both the traces. Solid line is a fit to a single exponential:  $k_{\text{obs}} = 1.5 \text{ s}^{-1}$ ;  $\%\Delta F = 4.6$ .

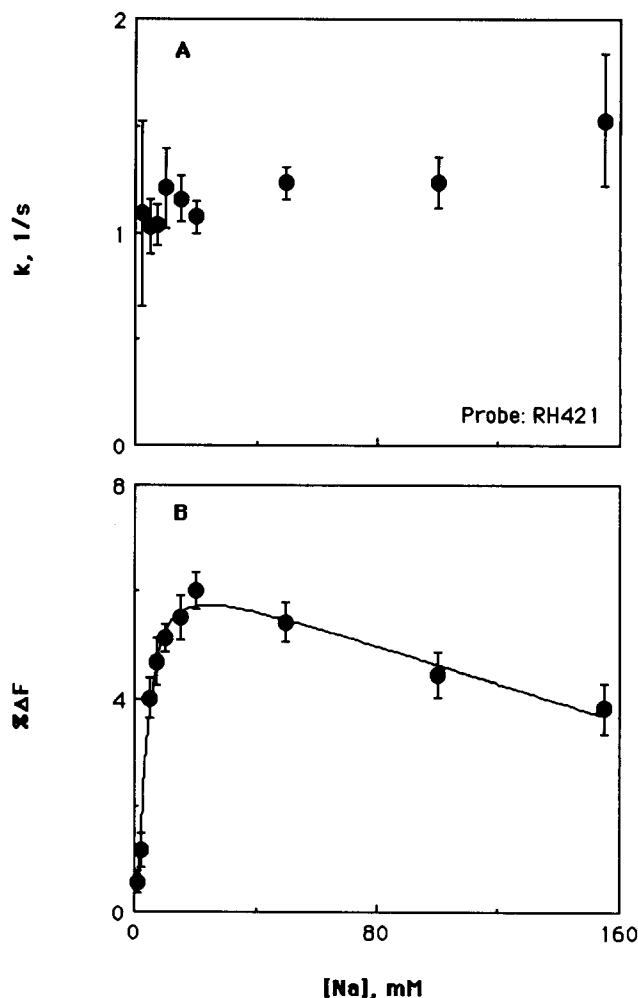


Fig. 6. Rate constant (A) and magnitude (B) of RH421-enzyme fluorescence change as a function of  $[\text{Na}^+]$  when  $\text{Mg}^{2+}$  is replaced by  $\text{Ca}^{2+}$ . Experiments were performed as in Fig. 2, except that  $\text{Mg}^{2+}$  was replaced with 4 mM  $\text{Ca}^{2+}$ . The solid line is a fit to the Hill equation for the rising part and a straight line for the falling part of the data ( $\%\Delta F_{\text{max}} = 6.4$ ;  $K_{1/2} = 4 \text{ mM}$ ;  $n_H = 1.8$ ).

with  $\text{Ca}^{2+}$  was similar to that of RH421 with  $\text{Mg}^{2+}$  in the presence of oligomycin. Further, unlike the case with  $\text{Mg}^{2+}$ , with  $\text{Ca}^{2+}$  oligomycin abolished the RH421 response to ATP (Fig. 5, lower trace). These results support possibility (b) above.

In the presence of  $\text{Ca}^{2+}$ ,  $k_{\text{obs}}$  for the RH421 fluorescence change did not vary with  $\text{Na}^+$  over the concentration range examined (1–155 mM; Fig. 6A). The magnitude of fluorescence change (Fig. 6B) varied in approximately the same manner as with  $\text{Mg}^{2+}$  in the presence of oligomycin (Fig. 2B, filled symbols):  $\%\Delta F$  increased at low  $[\text{Na}^+]$ , to a maximum at about 20 mM  $[\text{Na}^+]$ , and then decreased at higher  $[\text{Na}^+]$ . As with RH421 enzyme with  $\text{Mg}^{2+}$  + oligomycin, the data were fitted to a Hill equation for the rising phase plus a straight line for the declining phase; a minimum value for  $K_{1/2}$  (Fig. 6B, line), so estimated, was 4 mM.

### Effects of $Mn^{2+}$

$Mn^{2+}$  is partially effective in activating the  $Na^+/K^+$ -ATPase [30,31], and also favors  $E_2$  conformations [31]. Thus, in the presence of  $Mn^{2+}$  and in low  $[Na^+]$  ( $< 50$  mM), IAF-enzyme fluorescence corresponded to that of the  $E_2$  conformation. When enzyme under these conditions was mixed rapidly with 1 mM ATP, an initial increase in fluorescence was observed (Fig. 7A and Table I), attributable to the  $E_2 \rightarrow E_1 \cdot ATP$  transition. The observed rate-constant for this fluorescent increase ( $7.7\ s^{-1}$ ; Table I) was consistent with the rate-constant reported for the  $E_2(K) \rightarrow E_1$  transition [20]. (Under the conditions of this experiment, the enzyme can undergo a further transition from  $E_2$  to  $E_1P$ ; however, since the transition from  $E_1$  to  $E_1P$  is 'fluorescently silent', rate constants associated with this

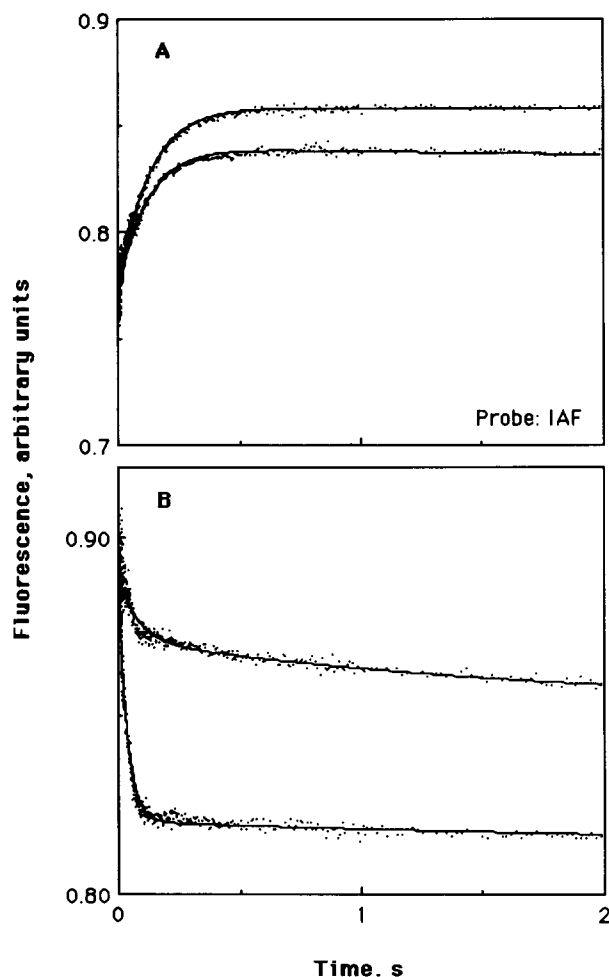


Fig. 7. Stopped-flow fluorescence response of IAF-enzyme when  $Mg^{2+}$  is replaced with  $Mn^{2+}$ . A. 100  $\mu\text{g/ml}$  enzyme, either untreated or with 10  $\mu\text{g/ml}$  oligomycin in buffer (25 mM imidazole-HCl (pH 7), 1 mM NaCl, 154 mM choline-Cl, 1 mM EDTA, 4 mM  $MnCl_2$ ) is rapidly mixed with 1 mM ATP in the same buffer. Untreated, upper trace; oligomycin-treated, lower trace. B. Same as in A, except that  $[NaCl] = 155$  mM,  $[choline-Cl] = 0$ . Solid lines are fits to two exponentials. Untreated, lower trace; oligomycin-treated, upper trace. Fit parameters are given in Table I.

TABLE I

Parameters of IAF fluorescence change when  $Mg^{2+}$  is replaced by  $Mn^{2+}$

Enzyme (100  $\mu\text{g/ml}$ ) in  $x$  mM  $Na^+$  (where  $x = 1$  or 155); 155- $x$  mM choline<sup>+</sup>; 4 mM  $Mn^{2+}$ ; 1 mM EDTA; 25 mM imidazole-HCl (pH 7), was rapidly mixed with 1 mM ATP in the same buffer (final concentrations: 50  $\mu\text{g/ml}$  enzyme, 0.5 mM ATP). In experiments with oligomycin, the enzyme was preincubated with 10  $\mu\text{g/ml}$  oligomycin. For each experiment, 7–9 shots were averaged under each condition; the averages were fitted to two exponentials. Traces of the experiments are shown in Fig. 7. The parameters reported are the averages of three experiments.

Condition	$k_1$	$\% \Delta F_1^*$	$k_2$	$\% \Delta F_2^*$
1 mM Na	$7.7 \pm 0.3$	$10.6 \pm 1.0$	$0.33 \pm 0.06$	$-0.5 \pm 0.2$
1 mM Na, + oligomycin	$8.0 \pm 0.3$	$8.6 \pm 1.4$	$0.39 \pm 0.03$	$-0.9 \pm 0.2$
155 mM Na	$29.8 \pm 0.8$	$-7.9 \pm 0.3$	$0.33 \pm 0.03$	$-0.3 \pm 0.3$
155 mM Na, + oligomycin	$12.4 \pm 1.0$	$-2.0 \pm 0.1$	$0.64 \pm 0.7$	$-2.0 \pm 0.2$

\* A positive  $\% \Delta F$  signifies an increase in fluorescence; a negative  $\% \Delta F$  signifies a decrease.

transition will not be reflected in the observed fluorescence change.)

With 1 mM  $Na^+$ , this initial increase in fluorescence (associated with the formation of  $E_1$  and  $E_1P$  species) was followed by a small decrease, attributable to the transition to  $E_2P$ . As the  $Na^+$  concentration was raised, both the magnitude and rate constant for this transition increased; however, with these intermediate concentrations, both the initial increase and the subsequent decrease in fluorescence occur, and it was thus difficult to evaluate each change separately.

At high  $[Na^+]$  ( $> 50$  mM), the initial fluorescence increase disappeared, indicating that at these  $[Na^+]$  the enzyme was initially in the  $E_1$  conformation even in the presence of  $Mn^{2+}$ ; then only a decrease in fluorescence was observed after mixing with ATP, reflecting a transition to  $E_2P$  (Fig. 7B). The rate-constant for this transition at 155 mM  $Na^+$  (Table I) was lower than that observed with  $Mg^{2+}$  ( $30\ s^{-1}$  with  $Mn^{2+}$  as opposed to  $80\ s^{-1}$  with  $Mg^{2+}$  [18]), and the magnitude of the fluorescent change was slightly larger than that with  $Mg^{2+}$  ( $-8\%$  with  $Mn^{2+}$  as opposed to  $-6\%$  with  $Mg^{2+}$  [18]).

In the presence of oligomycin, the initial increase in fluorescence disappeared at  $[Na^+] \geq 15$  mM. This greater sensitivity to  $Na^+$  is consistent with oligomycin favoring  $E_1$  conformations even in the absence of enzyme phosphorylation, as reported earlier [32]. Further, oligomycin reduced the rate constant for the fluorescent decrease at 155 mM  $Na^+$  by only 60% (from  $30\ s^{-1}$  to  $12\ s^{-1}$ ). By contrast, oligomycin in the presence of  $Mg^{2+}$  reduced  $k_{obs}$  by two orders of magnitude (from  $80\ s^{-1}$  to  $1\ s^{-1}$  [18]). Thus, the effect

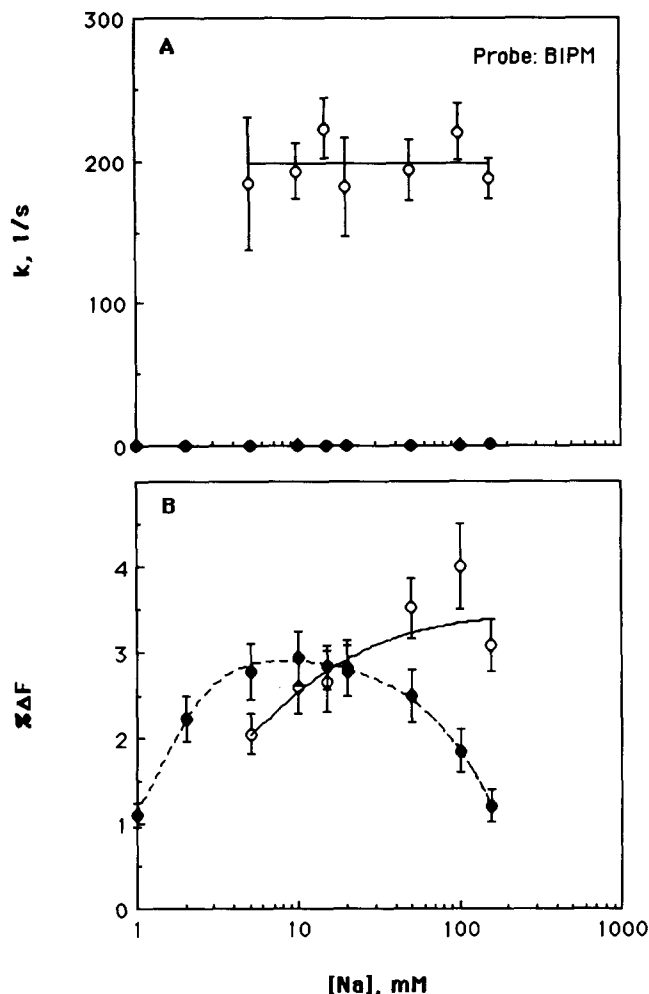


Fig. 8. Rate constant (A) and magnitude (B) of fluorescence change with BIPM-enzyme as a function of  $[Na^+]$  in the absence (○) and presence (●) of oligomycin. In panel A, the solid line represents the average  $k_{obs}$  ( $199 \pm 8 \text{ s}^{-1}$ ). In panel B, the solid line represents a fit to the Hill equation ( $\% \Delta F_{max} = 3.5 \pm 0.4$ ,  $K_{1/2} = 3.5 \pm 1.3$ , and  $n_H = 1.0 \pm 0.6$ ). The dashed line represents a fit to the Hill equation + a straight line ( $\% \Delta F_{max} = 3.0 \pm 0.2$ ,  $K_{1/2} = 1.3 \pm 0.1$ ,  $n_H = 2.2 \pm 0.7$ , and for the straight line, slope =  $-0.0118 \pm 0.0018$ ).

of oligomycin in stabilizing the  $E_1P(3Na^+)$  form of the enzyme seemed sensitive to the divalent cation used.

#### Fluorescence changes in enzyme labeled with BIPM

Taniguchi and co-workers [33] have reported that fluorescence changes in enzyme labeled with BIPM occur at several steps in the enzyme cycle, including conformational changes and  $Na^+$  binding and occlusion. We have examined changes in the kinetics of fluorescence change in BIPM-enzyme as a function of  $Na^+$  concentration, with and without oligomycin (Fig. 8).

In the absence of oligomycin,  $k_{obs}$  was  $199 \text{ s}^{-1}$  and largely independent of  $[Na^+]$  over the range examined (Fig. 8A). This value for  $k_{obs}$  was much higher than that for IAF and RH421 signals, and also much higher

than those reported by Taniguchi et al. [8]. The magnitude of the change also increased with a  $K_{1/2}$  of about 3.5 mM and a maximum value of 3.5%. This value of  $\% \Delta F_{max}$  is slightly higher than that observed by Taniguchi et al. [8].

In the presence of oligomycin,  $k_{obs}$  decreased by a factor of 200. The magnitude of the fluorescence change increased to a maximum value of 3.0%, as in the absence of oligomycin; however, this maximum was obtained at lower values of  $[Na^+]$  than in the absence of oligomycin (10 mM vs. 155 mM). As  $Na^+$  was increased further,  $\% \Delta F$  declined, so that at 155 mM  $Na^+$  the magnitude was about the same as at 1 mM  $Na^+$ . Such an increase and decrease in  $\% \Delta F$  was also observed in RH421 fluorescence response with oligomycin (Fig. 8B).

#### Discussion

Early experiments on the  $Na^+/K^+$ -ATPase identified two distinct conformations ' $E_1$ ' and ' $E_2$ ', distinguished in terms of numerous criteria [1–11]. In previous work [18], fluorescence changes of IAF-enzyme were followed as a function of  $[Na^+]$ . Assuming the IAF fluorescence change reflected the conformational change from  $E_1P$  to  $E_2P$ , those results indicated that the release of at least one  $Na^+$  preceded the conformational change, and that the rate-limiting step over the sequence examined was the release of the second and/or the third  $Na^+$ . But how valid is the assumption that the IAF fluorescent signal represents the conformational change? The related question is: do the properties of all indicators of conformational transitions change at the same step in the enzyme cycle, or is there a series of steps over which these change? From this question arises a fundamental question: are there just two conformations for the enzyme, or is there a spectrum of conformations that the enzyme goes through between the traditional  $E_1$  and  $E_2$  conformations? And: at what step in the enzyme cycle is charge transferred across the membrane?

The experiments presented here address these questions through comparing, under various conditions, the signals generated by two fluorescent indicators of conformational change and one indicator of charge transfer. From the data presented here, we conclude:

(i) The rate-constant ( $k_{obs}$ ) of the RH421 fluorescence change increased with  $[Na^+]$ , similar to that of IAF-fluorescence change [18]. The  $K_{1/2}$  for  $Na^+$  for both signals was approximately the same. Thus, assuming RH421 reports the transfer of charge across the membrane [26,27], the rate-limiting step for charge transfer was the same as for the IAF fluorescence change – the release of the second and/or the third  $Na^+$ . Nevertheless, the maximum  $k_{obs}$  for RH421 fluorescence change was consistently higher (by about 25%)

than the  $k_{\text{obs}}$  for the IAF fluorescence change ( $100 \text{ s}^{-1}$  for RH421 vs.  $80 \text{ s}^{-1}$  for IAF [18]), indicating that RH421 reports an earlier step in the enzyme cycle than IAF. A caveat to this conclusion is the possibility that RH421 increases the rate of release of the second and/or the third  $\text{Na}^+$ .

(ii) With oligomycin-treated enzyme,  $k_{\text{obs}}$  for the RH421 fluorescence change decreased by two orders of magnitude and was  $\text{Na}^+$ -independent over the concentration range studied;  $\% \Delta F_{\text{max}}$  did not change significantly, similar to the sensitivity of IAF fluorescence change to oligomycin. Further,  $\% \Delta F$  for RH421 (and BIPM; see below) increased to a maximum at 20 mM  $[\text{Na}^+]$  and then decreased, so  $\% \Delta F$  at 155 mM  $\text{Na}^+$  was approximately the same as that for 1 mM  $\text{Na}^+$ . These results indicate that treatment with oligomycin slows down some (earlier)  $\text{Na}^+$ -independent step so that this step then becomes rate-limiting. Moreover, the  $k_{\text{obs}}$  is consistent with the rate-constant for the dissociation of oligomycin from the enzyme [34]. Our observations are consistent with oligomycin inhibiting  $\text{Na}^+$  deocclusion, a step that is expected to be independent of  $[\text{Na}^+]$ , in accord with previous studies.

(iii) Oligomycin is thought to increase the affinity of the enzyme for  $\text{Na}^+$  [32,33]. Such an increased affinity would shift the equilibrium for the release of  $\text{Na}^+$  toward the sodium-bound forms ( $\text{EP} \cdot 3\text{Na}_{\text{out}}^+$ ,  $\text{EP} \cdot 2\text{Na}_{\text{out}}^+$ , and  $\text{EP} \cdot \text{Na}_{\text{out}}^+$ ). Under these conditions, the reaction sequence  $\text{EP}(3\text{Na}^+) \leftrightarrow \text{EP} \cdot 3\text{Na}_{\text{out}}^+ \leftrightarrow \text{EP} \cdot 2\text{Na}_{\text{out}}^+ \leftrightarrow \text{EP} \cdot \text{Na}_{\text{out}}^+ \leftrightarrow \text{EP}$  could be reversed at lower  $[\text{Na}^+]$  than in the absence of oligomycin. Thus, the decrease in fluorescence seen at higher  $[\text{Na}^+]$  in the presence of oligomycin could be attributed to diminished  $\text{Na}^+$  release from the enzyme.

(iv) When the total  $\text{Mg}^{2+}$  concentration was varied from 0.1 to 5 mM, the rate-constant of the RH421 fluorescence change ( $k_{\text{obs}}$ ) increased from  $10 \text{ s}^{-1}$  to  $90 \text{ s}^{-1}$  with a  $K_{1/2}$  of 0.22 mM. Analysis of these data in terms of  $[\text{MgATP}]$  and  $[\text{Mg}]_{\text{free}}$  yielded  $K_{1/2}$  values of 200  $\mu\text{M}$  and 13.2  $\mu\text{M}$ , respectively. It is not obvious which of these is functionally pertinent.

(v) When  $\text{Mg}^{2+}$  was replaced with  $\text{Ca}^{2+}$ , IAF fluorescence changes were small as well as slow. On the other hand,  $\% \Delta F$  for RH421-enzyme was 80% of that with  $\text{Mg}^{2+}$ , although  $k_{\text{obs}}$  was reduced by a factor of 100. These observations support our conclusion [(i) above] that RH421 reports a step in the enzyme cycle prior to that reported by IAF. Oligomycin abolished the RH421 fluorescence change with  $\text{Ca}^{2+}$ , indicating that  $\text{Ca}^{2+}$  affects the enzyme cycle at two steps: (a) the step reported by IAF, which is largely blocked; and (b) the step that is also blocked by oligomycin. Since oligomycin and  $\text{Ca}^{2+}$  each reduced the rate-constant by a factor of 100, then if these reductions occurred at different steps, the rate-constant with oligomycin plus  $\text{Ca}^{2+}$  would be the same as the rate-constant with

either oligomycin or  $\text{Ca}^{2+}$ ; however, if the reductions were at the same step, then the factor by which the rate-constant was reduced in the presence of oligomycin plus  $\text{Ca}^{2+}$  would be a product of the factors by which the rate-constant were reduced with either alone (i.e., a factor of  $100 \times 100$ ); these signals would be too slow to be observed in the time-scale that these experiments were performed, and the fluorescence would appear to be unchanged. This was observed.

(vi) When  $\text{Mn}^{2+}$  was substituted for  $\text{Mg}^{2+}$ , the enzyme was initially in the  $\text{E}_2$  conformation, as indicated by the steady-state level of IAF fluorescence. This low level of fluorescence was maintained with  $[\text{Na}^+] < 50 \text{ mM}$ . When the  $\text{Mn}^{2+}$ -enzyme was then rapidly mixed with ATP, the IAF fluorescence increased, indicating a transition to the  $\text{E}_1$  conformations. The rate constant for this conformational change was comparable to that observed by Steinberg and Karlsh for the  $\text{E}_2(\text{K}) \rightarrow \text{E}_1 \cdot \text{ATP}$  conformational change, indicating that this fluorescence change reflected an  $\text{E}_2(\text{Mn}^{2+}) \rightarrow \text{E}_1 \cdot \text{ATP}$  transition.

The ATP-induced changes in the presence of  $\text{Mn}^{2+}$  and either 1 or 155 mM  $\text{Na}^+$  (Table I) were measured with the ionic strength kept constant with choline chloride. ATP-induced changes have also been measured earlier [18] with enzyme plus  $\text{Mg}^{2+}$  in low ionic-strength buffer ( $\text{NaCl} \leq 5 \text{ mM}$ , 25 mM imidazole-HCl (pH 7)) in the absence of  $\text{K}^+$ ; under these conditions, the enzyme underwent a rapid transition from the  $\text{E}_2$  to the  $\text{E}_1$  conformation, as indicated by a rapid increase in fluorescence ( $k_{\text{obs}} > 100 \text{ s}^{-1}$ ). The latter rate constant was much higher than that for the fluorescence increases seen with  $\text{Mn}^{2+}$  at low  $\text{Na}^+$  (this work) and in the presence of  $\text{K}^+$  [20]. These observations imply either (a) the fluorescence change indicates a step that occurs after  $\text{K}^+$  deocclusion, as proposed by Skou and Esmann [9], or (b) the pathway for conformational transition from  $\text{E}_2 \rightarrow \text{E}_1 \cdot \text{ATP}$  is different in the presence or absence of  $\text{K}^+$  (or  $\text{Mn}^{2+}$ ).

(vii) The rate-constant of the fluorescence change with BIPM-labelled enzyme was about two-fold greater than that with IAF-enzyme or with RH421, and independent of  $[\text{Na}^+]$  over the range examined (1–155 mM). Thus BIPM reported a step prior to those reported by IAF or RH421 (i.e., the release of the second and/or third  $\text{Na}^+$ ). Under the conditions of our experiments ( $[\text{Na}^+] = 1\text{--}155 \text{ mM}$ ), the release of the first  $\text{Na}^+$  was practically irreversible, so this step could be the one reported by BIPM. This possibility is strengthened by the observation that DCCD, which blocks  $\text{Na}^+$  occlusion [36], binds close to the BIPM binding site (Glu-953 for DCCD, vs. Cys-964 for BIPM [37]).

Note that  $k_{\text{obs}}$  at 155 mM  $\text{Na}^+$  was much greater than the rate-constant observed by Taniguchi et al. under similar conditions (160 mM  $\text{Na}^+$ ; [8]): approxi-



mately  $200 \text{ s}^{-1}$  vs.  $30\text{--}35 \text{ s}^{-1}$ . Those low rate constants are inconsistent with the rate-constants observed with IAF- and RH421-enzyme. Possible explanations for this discrepancy are that Taniguchi et al. [8] did most of their experiments at high ionic strength ( $0.6\text{--}2 \text{ M}$ ), or that they used a much lower concentration of ATP ( $5 \mu\text{M}$ ). The latter possibility is particularly appealing because Steinberg and Karlish [20] have reported that  $k_{\text{obs}}$  for IAF-enzyme increases with [ATP] with a  $K_{1/2}$  of approx.  $50 \mu\text{M}$ . In fact, Taniguchi et al. [38] found that at  $2 \text{ M Na}^+$ , they could not saturate the rate-constant for BIPM fluorescence change with  $300 \mu\text{M}$  ATP, although  $\% \Delta F$  and the amount of EP formed were saturated at a much lower [ATP].

Because oligomycin reduced  $k_{\text{obs}}$  but not  $\% \Delta F_{\text{max}}$ , the BIPM fluorescence change represents a step that is either irreversible or is separated from the oligomycin-sensitive step by an irreversible step. Again, this conclusion is consistent with BIPM reporting the release of the first  $\text{Na}^+$ .

(viii) The rate constant for fluorescence change of BIPM-enzyme incubated with oligomycin was comparable to that of enzyme labelled with IAF and RH421 with oligomycin. Thus, the oligomycin-inhibited step occurs at or before the step reported by BIPM. Since oligomycin inhibits the deocclusion of  $\text{Na}^+$ , the step reported by BIPM is either  $\text{Na}^+$  deocclusion or the release of the first  $\text{Na}^+$ .

(ix) The magnitude of the fluorescence change with BIPM increased with  $\text{Na}^+$  to a maximum of about 3.5% and a  $K_{1/2}$  of 3.5 mM. In the presence of oligomycin, this curve shifted to lower  $\text{Na}^+$  concentrations – the  $K_{1/2}$  was now 1.4 mM; however,  $\% \Delta F_{\text{max}}$  remained the same. This implies that the step reported by BIPM is separated from the rate limiting step in the presence of oligomycin ( $\text{Na}^+$  deocclusion) by an irreversible step. Therefore, these results support the conclusion that BIPM reports the release of the first  $\text{Na}^+$  from the enzyme.

The decrease in  $\% \Delta F$  at higher  $[\text{Na}^+]$  (seen also with RH421-enzyme treated with oligomycin) reflects the reversal of the enzyme cycle from E(high BIPM fluorescence) to E(low BIPM fluorescence), consistent with oligomycin preventing the deocclusion of  $\text{Na}^+$ . Such a shift should be observed even in the absence of oligomycin, but at much higher  $[\text{Na}^+]$ , and has been observed with IAF-enzyme [20].

(x) From the observations presented here, we have constructed a scheme (Fig. 9) for the transition of the enzyme cycle with emphasis on the portion involved in  $\text{Na}^+$  (and charge) transport, indicating the steps that we believe are reported by IAF, BIPM and RH421, and the steps inhibited by oligomycin and  $\text{Ca}^{2+}$ . This scheme does not refer to the  $E_1$  and  $E_2$  conformations; instead, it considers each step in the enzyme sequence to be a distinct conformation with  $\text{Na}^+$  bound to the

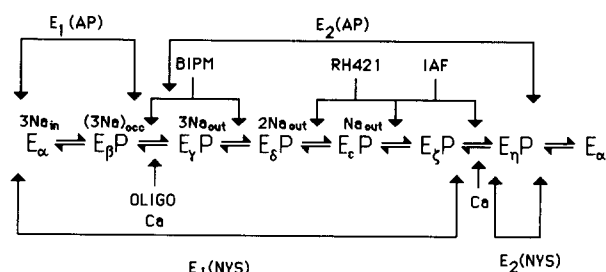


Fig. 9. Suggested model for the action of the  $\text{Na}^+/\text{K}^+$ -ATPase based on the results with IAF, RH421 and BIPM. The Greek letters represent different conformations; the steps affected by oligomycin and  $\text{Ca}^{2+}$  are shown. The proposed steps indicated by IAF, RH421, and BIPM are indicated.  $E_1(\text{AP})$  and  $E_2(\text{AP})$  represent the species in this scheme that correspond to the  $E_1$  and  $E_2$  conformations proposed by Albers and Post;  $E_1(\text{NYS})$  and  $E_2(\text{NYS})$  represent the assigned conformations in the Nørby-Yoda-Skou scheme. The subscripts 'in' and 'out' indicate that the bound  $\text{Na}^+$  is in equilibrium with intracellular or extracellular  $\text{Na}^+$ , respectively; 'occ' represents occluded  $\text{Na}^+$ .

enzyme as shown. Subscripts indicate the side of the membrane from which the cation binding sites are accessible; occlusion is further indicated by enclosing  $\text{Na}^+$  within parentheses.

We have indicated also which of the species in the scheme would be identified as ' $E_1$ ' and ' $E_2$ ' according to the Albers-Post (AP) and the Nørby-Yoda-Skou (NYS) schemes [18]; for the latter scheme, following Skou's proposal [17], we have included two EP forms with no  $\text{Na}^+$  bound ( $E_\gamma\text{P}$  and  $E_\eta\text{P}$ , corresponding to EP with three and two cation binding sites, respectively), although a change from three to two cation sites (and thus conversion of ' $E_1\text{P}$ ' to ' $E_2\text{P}$ ') could also be placed between  $E_\gamma\text{P}$  and  $E_\delta\text{P}$ .

(xi) We believe that the analysis of transient kinetic data must de-emphasize the traditional interpretation of  $\text{Na}^+/\text{K}^+$ -ATPase kinetics in terms of two conformations. We believe that such interpretations are prone to misleading and contradictory conclusions. A similar point was raised by Glynn and Karlish [39], who pointed out that the nomenclature in terms of  $E_1$  and  $E_2$  is insufficient for explaining many experimental results. Nørby [40] suggested that a hierarchy of conformational states be considered, with transitions between subconformations within each hierarchical level. Jencks [41] has raised similar arguments for the  $\text{Ca}^{2+}$ -ATPase from the sarcoplasmic reticulum.

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